

Effect of the addition of t-ZrO₂ on the material properties of β -TCP/PCL composites

Ho-Yeon Song · Do Van Quang · Young-Ki Min ·
Byong-Taek Lee

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Abstract Using poly-methyl methacrylate as a pore-forming agent, porous β -tricalcium phosphate (β -TCP)/t-ZrO₂ composites were fabricated depending on the volume percentages (vol.%) of t-ZrO₂ powder. In the porous sintered bodies, hybrid pores, about 20 and 200 μ m in diameter, were homogeneously dispersed in the β -TCP/t-ZrO₂ matrix and showed good interconnection. On the other hand, β -TCP-(t-ZrO₂)/polycaprolactone (PCL) composites were fabricated by the melt infiltration process using porous β -TCP/t-ZrO₂ bodies. The relative density of the β -TCP-(t-ZrO₂)/PCL composites increased as the vol.% of t-ZrO₂ increased and its maximum value was about 98.6%. However, the hardness, bending strength and elastic modulus of β -TCP-(t-ZrO₂)/PCL composites decreased due to the low densification of porous β -TCP/t-ZrO₂ bodies as the volume percentages of t-ZrO₂ content increased. The values (using 20 vol.% of t-ZrO₂) were 11.4 Hv, 25.5 MPa and 17.2 GPa, respectively.

Introduction

β -Tricalcium phosphates (β -Ca₃(PO₄)₂, β -TCP) have been widely used as biomaterials for bone substitutes [1–3] because their chemical composition is similar to that of human bone and teeth. Moreover, β -TCP has some unique properties such as excellent biocompatibility and osteo-integration [4]. Its high potential for resorption in the human biological environment allows a significant recovery of the

operational site by advancing bone growth during the progressive degradation of this material [5]. However, β -TCP has poor mechanical properties, such as low fracture strength and low fracture toughness compared to human bone. On the other hand, polycaprolactone (PCL) has been used for bone substitutes and as a drug carrier because of its typical biodegradable property with excellent biocompatibility [6, 7].

In previous work, BCP/PCL composites were successfully fabricated by infiltration of molten PCL into porous BCP bodies, using 70 vol.% of poly-methyl methacrylate (PMMA) powder as a pore-forming agent (about 200 μ m in diameter). However, when the percentage of PMMA powder was increased to 80 vol.%, the porous bodies were quite weak and broke easily during the infiltration process of molten PCL. Therefore, to make sound β -TCP-(t-ZrO₂)/PCL composites, porous β -TCP bodies using 80 vol.% of PMMA powder should be reinforced to keep their original shape during the infiltration process. Recently, yttria (3 mol%) stabilized zirconia (t-ZrO₂) has received a lot of attention for artificial implants due to its biocompatibility and corrosion resistance, high fracture toughness, excellent chemical stability and outstanding mechanical properties [8–10]. Hence, t-ZrO₂ is considered to be one of the most suitable reinforcements for porous β -TCP bodies.

In this work, to make porous β -TCP/t-ZrO₂ bodies, hybrid PMMA powders (20 and 200 μ m in diameter) were used as a pore-forming agent. Then, using the melt infiltration process, the molten PCL was infiltrated into the porous β -TCP/t-ZrO₂ bodies to obtain the β -TCP-(t-ZrO₂)/PCL composites.

Experimental procedure

As starting materials, biphasic calcium phosphate (BCP) powders (about 80 nm) were synthesized by a microwave-

H.-Y. Song · D. Van Quang · Y.-K. Min · B.-T. Lee (✉)
Department of Biomedical Engineering and Materials,
School of Medicine, Soonchunhyang University, 366-1,
Ssangyoung-dong, Cheonan, Chungnam 330-090, South Korea
e-mail: lbt@sch.ac.kr

assisted process [11]. Spherical t-ZrO₂ powder (about 70 nm, Tosoh, Japan) and PMMA powders (about 20 and 200 μm in diameter, LG chemical company, Korea) as a pore-forming agent were used. First, BCP and different volume percentages (20, 30, 40, 50 vol.%) of t-ZrO₂ powders were mixed for 24 h by wet-ball milling using Al₂O₃ balls. Then, the mixture powders were dried on a hot plate while stirring. The BCP/t-ZrO₂ powders and 80 vol.% of PMMA powders (20 and 200 μm in diameter) were mixed for 15 h by dry-ball milling also using Al₂O₃ balls. The mixture powders after dry-ball milling were compacted into pellets by a uni-axial press and were burned out in an air atmosphere at 700 °C. Finally, the burn-out samples were sintered at 1,400 °C in an air atmosphere.

On the other hand, PCL (Aldrich, USA) was melted at 120 °C, and then poured into a mold containing porous β-TCP/t-ZrO₂ bodies. They were kept at the same melt temperature of the PCL for about 5–10 min before natural cooling. After complete solidification of the melted PCL, the PCL on the outside of the samples was removed to take β-TCP-(t-ZrO₂)/PCL composites.

The relative density of β-TCP-(t-ZrO₂)/PCL composites was measured by the Archimedes method. The pore size and microstructure of the porous β-TCP/t-ZrO₂ bodies and β-TCP-(t-ZrO₂)/PCL composites were investigated by scanning electron microscopy (SEM, JSM-635F, Jeol). To identify the crystal structure and phases, an X-ray diffractometer (XRD, D/MAX-250, Rigaku, Japan) were used. The hardness was measured using a Vickers hardness tester (Hv-112, Akashi, Japan) by indentation with a load of 0.3 kg (10 points/sample). For bending strength and elastic modulus measurements, the samples were bar-shaped (4 × 4 × 35 mm³). The bending strength measurement was carried out using a four-point bending method with a span of 10 mm and a crosshead speed of 0.1 mm/min, using a universal testing machine (UnitechTM, R&B, Korea). The elastic modulus was calculated using the following equation, which was developed by Grindosonic (J. W. Lemmens, MKS) [12].

$$E_R = 0.9465 * \frac{M * f^2}{w} * \frac{L_T^3}{f^3} * \left[1 + 6.59 * \left(\frac{t^2}{L_T^2} \right) \right]$$

where E_R is the elastic modulus, M is the weight, f is the frequency, w is the width, L_T is the length and t is the thickness of the specimen. When the elastic modulus was calculated, average values (15 times) were used.

Results and discussion

Figure 1 shows a SEM micrograph of (a) raw BCP, a TEM micrograph of (b) raw t-ZrO₂ powders, and SEM micrographs of two types of (c) and (d) raw PMMA powders. The particle size of the raw BCP powder was in the range

of 70–90 nm in diameter and the particle size of the spherical t-ZrO₂ powder was about 70 nm in diameter. Two types of PMMA powders, which were used together as a pore-forming agent, were spherical-shaped and about 20 and 200 μm in diameter. However, a few small particles of less than 10 μm in diameter were also observed.

Figure 2 shows SEM micrographs of porous β-TCP/t-ZrO₂ bodies depending on the t-ZrO₂ content. In Fig. 2a, containing 50 vol.% t-ZrO₂ powder, large amounts of special homogeneous pores about 200 μm in diameter were observed. From the enlarged image seen in Fig. 2b, it is confirmed that hybrid-type pores were formed in the porous β-TCP/t-ZrO₂ body due to the burn-out of the large (200 μm) and small (20 μm) PMMA powders located at the pore frame. On the surface of the large sized pores and their pore frames, many fine, interconnected pores were observed, which will have beneficial effects on the infiltration of the PCL. On the other hand, Fig. 2c and d show enlarged images of pore frame of porous β-TCP/t-ZrO₂ bodies using 20 and 50 vol.% of t-ZrO₂ powders, respectively. The t-ZrO₂ grains were less than 0.7 μm in diameter and were homogeneously dispersed in the β-TCP matrix. However, an important observation was that in the sample containing 20 vol.% t-ZrO₂ powder, a dense microstructure was found, whereas in the sample containing 50 vol.% t-ZrO₂ powder, some fine pores were observed as indicated with arrowheads. This means that the sinterability of porous β-TCP/t-ZrO₂ bodies decreased as the t-ZrO₂ content increased.

Figure 3 shows XRD profiles of (a) burn-out at 700 °C, sintered at 1,400 °C of β-TCP/t-ZrO₂ bodies containing (b) 20 vol.% of t-ZrO₂ and (c) 50 vol.% of t-ZrO₂ powders. In the burned-out sample, BCP containing hydroxyapatite (HAp) and β-TCP phases were detected as well as a t-ZrO₂ phase. However, in the sintered samples, neither HAp nor new reaction phases were found after the sintering process. The intensity of the t-ZrO₂ phase also slightly increased while the β-TCP phase slightly decreased as the t-ZrO₂ content increased.

Figure 4 shows the relative density, hardness, bending strength and elastic modulus of the β-TCP-(t-ZrO₂)/PCL composites depending on the t-ZrO₂ content. The relative density increased as the t-ZrO₂ content increased, and its maximum value was about 98.6%. However, the hardness, bending strength and elastic modulus of the β-TCP-(t-ZrO₂)/PCL composites slightly decreased as the t-ZrO₂ content increased. Using 20 vol.% of t-ZrO₂ content, the values were about 11.4 Hv, 25.5 MPa and 17.2 GPa, respectively. The value of the elastic modulus was controlled in the range of that of cortical bone [13]. As mentioned in Fig. 2d, some fine pores were observed when the t-ZrO₂ content increased. Therefore, after the infiltration process, the fine pores were also infiltrated by PCL,

Fig. 1 SEM and TEM micrographs of raw (a) BCP, (b) $t\text{-ZrO}_2$, (c) PMMA (200 μm) and (d) PMMA (20 μm) powders

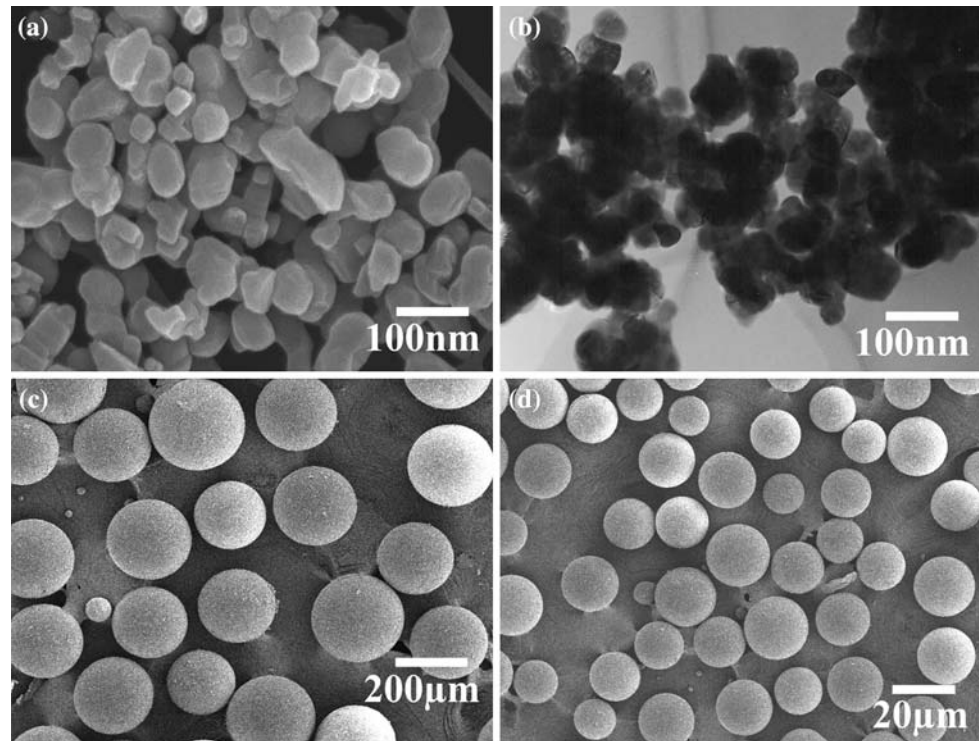
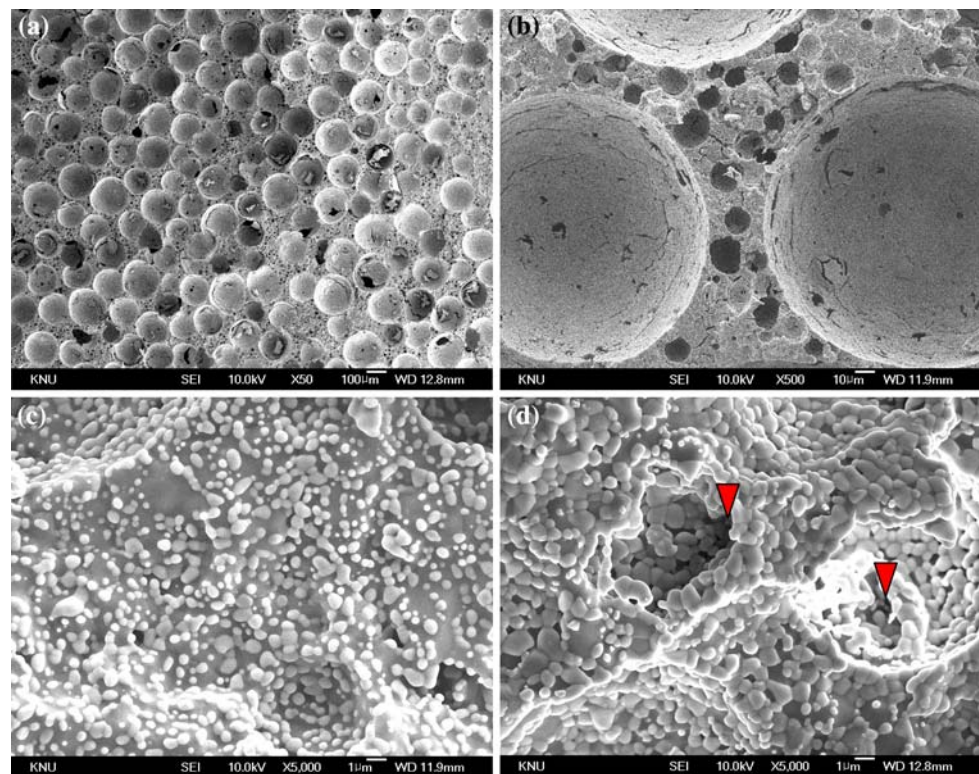


Fig. 2 SEM micrographs of porous $\beta\text{-TCP-(t-ZrO}_2\text{)}$ bodies using 50 vol.% of $t\text{-ZrO}_2$ (a) low magnification, (b) enlarged pore frame, high magnification of pore frame of porous $\beta\text{-TCP-(t-ZrO}_2\text{)}$ bodies using (c) 20 vol.% of $t\text{-ZrO}_2$ and (d) 50 vol.% of $t\text{-ZrO}_2$



which explained why the value of the relative density of the composites increased while the values of hardness, bending strength and elastic modulus of the composites slightly decreased as the vol.% of $t\text{-ZrO}_2$ increased.

Figure 5a and b show SEM micrographs and SEM fracture surfaces of $\beta\text{-TCP-(t-ZrO}_2\text{)}/\text{PCL}$ composites containing 20 vol.% $t\text{-ZrO}_2$, whereas Fig. 5c–e contain 50 vol.% $t\text{-ZrO}_2$ powders. In Fig. 5a, the dark contrast regions appear

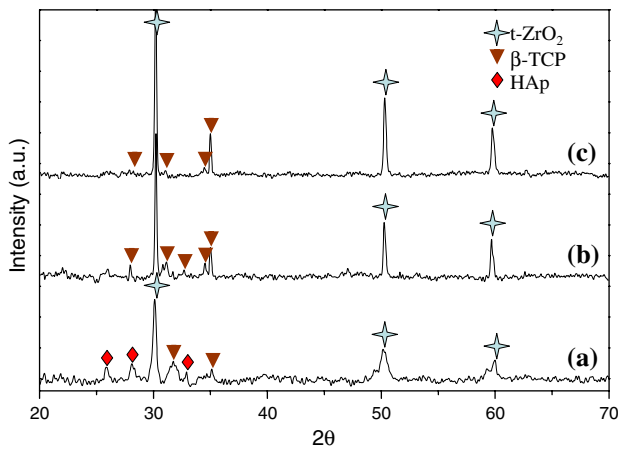


Fig. 3 XRD profiles of (a) burn-out at 700 °C, sintering at 1,400 °C of β -TCP-(t-ZrO₂) bodies containing (b) 20 vol.% of t-ZrO₂ and (c) 50 vol.% of t-ZrO₂

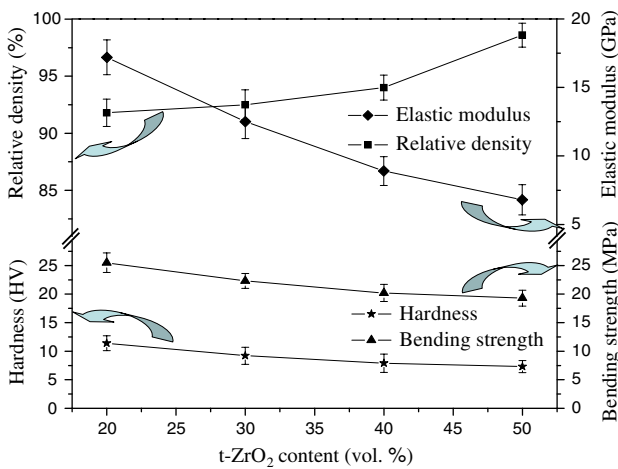
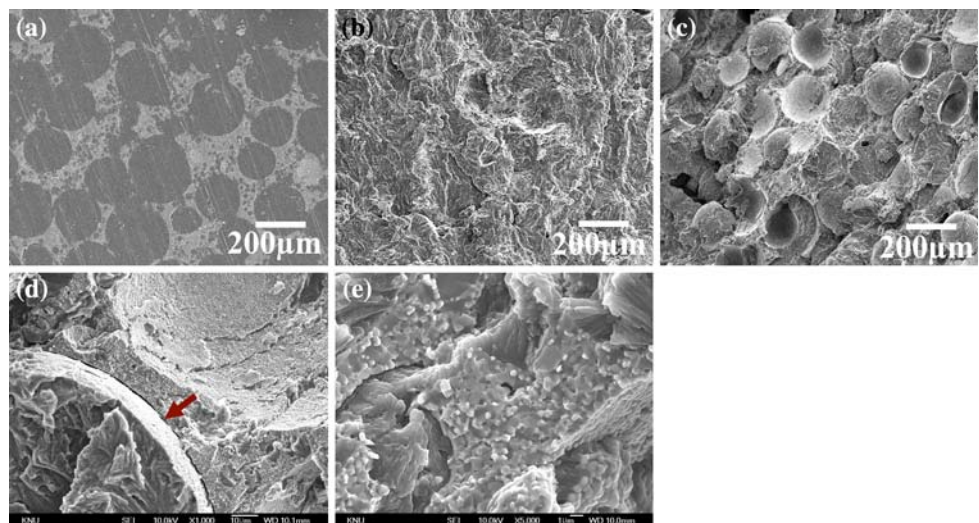


Fig. 4 Relative density, elastic modulus, hardness and bending strength of β -TCP-(t-ZrO₂)/PCL composites depending on the t-ZrO₂ content

Fig. 5 SEM micrographs of β -TCP-(t-ZrO₂)/PCL composites using 20 vol.% of t-ZrO₂ (a) polished surface, (b) fracture surface, (c) fracture surface of β -TCP-(t-ZrO₂)/PCL composites using 50 vol.% of t-ZrO₂, (d) enlarged image of (c) and (e) high magnification of pore frame of (c)



with a special shape, which corresponds to the infiltrated PCL phase while the bright regions are β -TCP-(t-ZrO₂) phases. Due to full infiltration of the PCL, the residual pores were hardly observed on the polished sample. In Fig. 5b, the fracture surface was flat and the pulled-out spherical PCL was hardly observed. This observation reveals that the interfaces between β -TCP-(t-ZrO₂)/PCL show comparatively strong bonding. In Fig. 5c, many pulled-out PCL spheres and their traces were observed on the fracture surface of the β -TCP-(t-ZrO₂)/PCL composites although some of the PCL spheres were torn during the fracture. This observation means that the interfacial strength between the β -TCP/t-ZrO₂ frame and PCL spheres was low compared with that of the PCL. In Fig. 5d, which is an enlarged image of Fig. 5c, the gap between the PCL and pore frame was clearly observed, as indicated with an arrow. Figure 5e shows high magnification of the pore frame of Fig. 5c, which is a typical mixed fracture mode, with trans-granular and inter-granular types that frequently appear in brittle ceramics.

Conclusion

Porous β -TCP/t-ZrO₂ bodies were fabricated depending on the vol.% of t-ZrO₂ (20, 30, 40, 50) using 80 vol. % of PMMA as a pore-forming agent. Due to the burn-out and sintering processes, the PMMA was completely removed. In the porous sintered bodies, the pores, about 20 and 200 μ m in diameter, were homogeneously dispersed in the β -TCP/t-ZrO₂ matrix and showed good interconnection between them. β -TCP-(t-ZrO₂)/PCL composites were fabricated by the melt infiltration process using porous β -TCP/t-ZrO₂ bodies and molten PCL. Most of the β -TCP/t-ZrO₂ pores were fully infiltrated by the molten PCL. As the volume percentage of t-ZrO₂ increased, many pulled-out PCL spheres and their traces were observed on the fracture

surface. The value of relative density of β -TCP-(t-ZrO₂)/PCL composites increased as the vol.% of t-ZrO₂ increased. However, the values of hardness, bending strength and elastic modulus of β -TCP-(t-ZrO₂)/PCL composites decreased as the vol.% of t-ZrO₂ content increased.

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